## GUAIANOLIDES FROM SAUSSUREA AFFINIS

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**Abstract**—Cynaropicrin,  $11\beta H$ -11,13-dihydrodesacylcynaropicrin, aguerins A and B, isoamberboin and the new guaianolides saussureolide and  $11\beta H$ -11,13-dihydrodesacylcynaropicrin 8- $\beta$ -D-glucoside were isolated from Saussurea affinis.

#### INTRODUCTION

Commercially available costus root oil from Saussurea lappa yields the germacradienolide costunolide and transformation products belonging to the elemadienolide, eudesmanolide and guaianolide series [1-3] whereas the above-ground parts of a few other representatives of the large genus Saussurea have furnished primarily a group of guaianolides related to cynaropicrin (1a) [1,4-9]. We

now report the results of our examination of Saussurea affinis whose range extends from north-east India to the Far East. The compounds isolated were cynaropicrin (1a), which is widely distributed in Cardueae [16], diol 2a previously isolated from Tricholepis glaberrima [10]‡, aguerin A (1b) and aguerin B (1c) which are found in several Centaurea species [11-14] and in Cousinia onopordioides [15], isoamberboin (3a) previously isolated from Jurinea maxima [17], and the new guaianolides 2b and 4a.

<sup>‡</sup>The C-11 epimer of **2a** has recently been isolated from Ainsliaea fragrans [18].

4a R-H

4b R = Ac

HINOR

···OR

30 R=H

36 R = Ac

# RESULTS AND DISCUSSION

One very polar new lactone was a  $\beta$ -D-glucoside ( $J_{1',2'}$  = 7.5 Hz) for which structure **2b** could be postulated on the basis of the <sup>1</sup>H NMR spectra of its TMS ether **2c** and its penta-acetate, **2d** (Table 1). The entire sequence of protons attached to the guaianolide carbon skeleton was established by extensive spin decoupling. Enzymatic hydrolysis furnished **2a** [10] and glucose. The glucose moiety was attached to the oxygen function on C-8 and not to that on C-3 because of the pronounced chemical shift of the H-3 signal to lower field on conversion of **2b** to **2d** 

A second new somewhat less polar substance was the tetrol, 4a, which we have named saussureolide. Comparison of the <sup>1</sup>H NMR spectrum of 4a with that of its triacetate 4b and extensive decoupling studies on the latter showed that the carbon skeleton and stereochemistry at the relevant centers were identical with those of its congeners 1a-1c and 2a. However, in the new compound the exocyclic methylene on C-4 was replaced by -CH<sub>2</sub>OH which, because of the multiplicities of the signals involved, had to be attached to a quaternary carbon. The presence of a vicinal tertiary hydroxyl on C-4 was inferred from the mass spectrum and the IR spectrum of 4b which continued to display hydroxyl absorption, and was confirmed by the <sup>13</sup>C NMR spectrum (Table 2). The stereochemistry at C-4 was deduced by examination of the <sup>1</sup>H NMR spectrum of 4b after reaction with trichloroacetylisocyanide (TAI) [19]; in the spectrum of the resulting monocarbamate (Table 1) the signals of H-3 and H-5 had moved to considerably lower field. Hence, the tertiary hydroxyl was cis to H-3 and H-5 and  $\alpha$ .

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Table 1. <sup>1</sup>H NMR spectra\* of compounds 2c, 2d, 4a and 4b

H No.	2c	2d	4a §	4b	4b + TA
1	2.82 br q (9)	2.98 br q	3.06 br q	3.21 br q	3.30 br q
2a	2.08 ddd (12, 8, 7)	2.43 m	2.19 ddd (14, 10, 7.5)	2.32 ddd	2.47 m
2b	1.66 ddd (12, 8, 7)	1.77 ddd	1.81 ddd (14, 10, 7.5)	1.78 ddd	1.91 ddd
3	4.43 br t (7)	5.53 br t	4.10 t (7.5)	5.07 t	5.81 t
5	2.73 br t (7)	2.80 br t	2.26 t (10.5)	2.43 t	3.44 t
6	4.01 t (10)	3.98 dd (10,9)	4.21 dd (10.5, 9.5)	4.23 dd	4.33 dd
7	2.20 q (10)	2.20  q	1.90  q  (10)	2.17  q	2.23 m
8	3.71 ddd (10, 8, 5)	3.66 ddd	3.63 ddd (10, 8, 5)	4.88 ddd	4.94 ddd
9a	2.66 dd (14, 5)	2.78 dd	2.77 dd (13, 5)	2.72 dd	2.68 dd
9b	2.37 dd (14, 8)	2.43 m	2.07 dd (13, 8)	2.13 dd	2.23 m
11	2.48 dq (10.5, 7)	2.43 m	2.53 dq (10, 7)	2.46 da	2.47 m
13†	1.43 d (7)	1.31 d	1.38 d (7)	1.23 d	1.30 d
14a	5.01 <i>br</i>	5.05 br‡	5.07 br	5.11 br	5.17 br
14b	4.98 br		5.01 br	5.09 br	5.09 br
15a	5.36 br	5.42 br	3.87	4.36	4.72
15b	5.36 br	5.29 br			
OAc†		2.09, 2.06	18000111	2.09, 2.08	2.12, 2.11
		2.03, 2.02		2.06	2.06
		2.00			
1′	4.34 d (7.5)	4.74 d			8.45 (NH)
2'	3.23 t (8)	5.00 t	_	atomic .	
3′	3.37 t (8)	5.21 t		*****	
4′	3.44 t (9)	5.07 t	_		
5′	3.21 dt (4.5, 9)	3.76 dt			
5′	3.71	4.19			

<sup>\*</sup>Run in CDCl<sub>3</sub> at 270 MHz with TMS as int. standard. Coupling constants (in parentheses) in Hz.

|Center of AB system.

The sesquiterpene lactone constituents of *S. affinis* are, thus, not significantly different from those previously found in other *Saussurea* species and, in fact, from those generally found in Cardueae which typically elaborate a series of related germacradienolides and guaianolides.

### EXPERIMENTAL

Isolation of S. affinis constituents. Above-ground parts of Saussurea affinis Spreng. (2 kg) collected in the vicinity of Arunachal Pradesh, India (voucher on deposite in herbarium of RRL, Jorhat), were extracted with CHCl3 in a Soxhlet apparatus until the extract was colorless. After removal of solvent at red. pres. the residue (71 g) was dissolved in 300 ml MeOH containing 10% H<sub>2</sub>O, allowed to stand overnight and filtered. The filtrate was washed with petrol (60–80°) (6  $\times$  300 ml), the MeOH portion was concd at red. pres. and the residue was thoroughly extracted with CHCl<sub>3</sub> (8 × 200 ml). Evaporation of the washed and dried extract furnished 27 g of a gummy residue which was chromatographed over 500 g of Si gel (60-120 mesh, BDH). Fractions (200 ml) were collected in the following order: 1-13 (C<sub>6</sub>H<sub>6</sub>), 114–113 ( $C_6H_6$ –EtOAc, 9:1), 114–121 ( $C_6H_6$ –EtOAc, 4:1), 122-179 (C<sub>6</sub>H<sub>6</sub>-EtOAc, 2:1), 180-193 (C<sub>6</sub>H<sub>6</sub>-EtOAc, 1:1), 194-249 (EtOAc), 250-269 (EtOAC-MeOH, 19:1), 270-275 (EtOAc-MeOH, 9:1), 276-280 (EtOAc-MeOH, 4:1) and 281 284 (EtOAc-MeOH, 2:1).

Fractions 58–74 (0.4g) which exhibited two major spots on TLC were combined and purified by prep. TLC ( $C_6H_6$ -EtOAc, 7:1). The faster moving band yielded 54 mg of aguerin B (1b) as a

gum; IR  $v_{\rm max}$  cm  $^{-1}$ : 3500, 1760, 1715, 1630, 1200, 1000 and 920;  $^{1}$ H NMR (270 MHz, CDCl<sub>3</sub>) as reported [14]; MS m/z 330 [M] $^{+}$ , 244, 226, 197, 69. Acetylation (Ac<sub>2</sub>O-pyridine overnight) provided 1e [11] which had the expected  $^{1}$ H NMR spectrum; IR  $v_{\rm max}^{\rm CHCl_3}$  cm  $^{-1}$ : 1760, 1710, 1630, 1225, 1150, 1000; MS m/z 372 [M] $^{+}$ , 330, 312, 244, 226, 197, 69. The slower moving band gave 45 mg of aguerin A (1c) [11]; IR  $v_{\rm max}$  cm  $^{-1}$ : 3500, 1760, 1720, 1630, 1200, 1035 and 915;  $^{1}$ H NMR similar to that of 1b except for substitution of the signals of the methacryl for those of the isobutyryl ester side chain; MS m/z 332 [M] $^{+}$ , 314, 262, 244, 226, 197, 71. Acetylation provided 1f which had the expected  $^{1}$ H NMR spectrum; IR  $v_{\rm max}^{\rm CHCl_3}$  cm  $^{-1}$ : 1760, 1725, 1630, 1220, 1150, 1000; MS m/z: 374 [M] $^{+}$ , 332, 244, 226, 197, 71.

Fractions 81-94 (120 mg), which exhibited one major spot, were combined and purified by prep. TLC (C<sub>6</sub>H<sub>6</sub>-EtOAc, 2:1) to yield 30 mg of impure (by NMR criteria) isoamberboin (3a) [17, 20–22] as a gum, <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>):  $\delta$ 5.06 and 4.76 (br, H-14ab,), 3.93 (t, J = 9 Hz, H-6), 3.75 (dt, J = 6, 9 Hz), 3.12m (H-1), 2.82 (dd, J = 13, 6 Hz, H-9a), 2.5 (c, H-11, H-5), 2.25 (c, H-11, H-5)9b, H-2a,b), 2.05 (q, J = 10 Hz, H-7), 1.44 (d, J = 7 Hz, H-13), 1.24 (d, J = 7 Hz, H-15); MS m/z 264 [M]<sup>+</sup>, 246, 218, 203. Acetylation (Ac<sub>2</sub>O-pyridine overnight) provided 3b (jurmolide) [17,23] which was pure but could not be induced to crystallize. IR  $v_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 1760, 1730 (broad), 1650, 1100, 950; MS m/z: 306 [M] $^{\frac{1}{4}}$ , 264, 246, 218, 203;  $^{1}$ H NMR (CDCl<sub>3</sub>):  $\delta$ 5.15 and 4.82 (*br*, H-14a,b), 4.93 (dt, J = 6, 9 Hz, H-8), 4.01 (t, J = 9 Hz, H-6), 3.11 (m, H-1), 2.89 (dd, J = 13, 6Hz, H-9a), 2.54 (t, J = 9Hz, H-5), 2.4 (c, H-11, H-2a), 2.24 (c, H-7, H-2b), 2.12 (Ac), 1.35 (d, J = 7 Hz, H-13), 1.24 (d, J = 15 Hz, H-15);  ${}^{1}$ H NMR ( $C_{6}D_{6}$ ) (de-

<sup>†</sup>Intensity of three protons.

<sup>‡</sup>Intensity of two protons.

<sup>§</sup>Three drops DMSO-d6 added.

Table 2. 13C NMR spectra\* of compounds 2d and 4b

C No.	2d	4b
1	44.37 d†	41.46 <i>d</i> †
2	36.34 t†	32.79 d†
3	74.67 d†	78.24 d†
4	147.89	81.39
5	50.91 d†	55.00 d†
6	78.92 d†	76.90 d
7	53.93 d†	53.66 d†
8	85.10 d†	75.43 d
9	42.55 t	42.03 t
10	142.56	140.49
11	41.46d	40.46 d
12	177.94	176.85
13	16.12 q	15.71 q
14	116.81 t	117.19 t
15	114.33 t	64.13 t
Ac-C-O	170.62, 170.42, 170.24	170.42, 170.31, 170.04
	169.33, 169.04	
Ac-Me	21.24, 20.70, 20.70	21.11, 21.11, 20.92
	20.57, 20.57	
1'	101.28 d	
2'	71.94 d	_
3'	73.08 d	
4'	68.46 d	_
5′	72.05 d	
6'	62.19 t	_

<sup>\*</sup>Run at 67.89 MHz in CDCl<sub>3</sub> with TMS as int. standard. Unmarked signals are singlets.

coupled to verify  $J_{7,11}$ ):  $\delta 2.14$  (dt, J=3, 9 Hz, H-1), 2.06 (dd, J=18, 3 Hz, H-2a), 1.9 (dd, J=18, 9 Hz, H-2b), 1.81 (m, H-4), 1.34 (m, H-5), 2.90 (t, J=9.5 Hz, H-6), 1.74 (q, J=10 Hz, H-7), 4.48 (dt, J=5, 10 Hz, H-8), 2.54 (dd, J=12, 5 Hz, H-9a), 1.63 (dd, J=12, 10 Hz, H-9b), 1.94 (m, H-11), 1.20 (d, J=7 Hz, H-13), 4.63 and 4.34 (br, H-14a,b), 1.20 (d, J=7 Hz, H-15), 1.62 (Ac). The value of  $J_{7,11}$  (10 Hz) and the solvent shift of the H-13 signal ( $\delta_{\text{CDCl}_3}$ - $\delta_{\text{C}_6}$ 06 + 0.15) indicated that the C-11 methyl group was pseudoequatorial and  $\alpha$  [24].

Fractions 117–119 (1.5 g) gave cynaropicrin (1a) identified by direct comparison (TLC, IR, NMR and MS) with an authentic sample [10]. Fractions 120–151 (480 mg) yielded, after purification by prep. TLC ( $C_6H_6$ –EtOAc, 2:1) 52 mg 2a identified by direct comparison with an authentic sample [10].

Fractions 210–224 (1.0g) were combined. Purification by prep. TLC (CHCl<sub>3</sub>–MeOH, 22:3) gave 120 mg of saussureolide (4a) as a gum, IR  $v_{\rm max}^{\rm Film}$  cm<sup>-1</sup>: 3500, 1770, 1050; MS m/z 298, 280, 267, 262, 249, 231. [Calcd for C<sub>15</sub>H<sub>22</sub>O<sub>6</sub>: MW, 298.1416. Found: MW(MS), 298.1419.] <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra are listed in Tables 1 and 2. Treatment with Ac<sub>2</sub>O–pyridine overnight and purification by prep. TLC provided the triacetate, 4b, whose <sup>1</sup>H NMR spectrum is listed in Table 1; IR  $v_{\rm max}^{\rm CHCl_3}$  cm<sup>-1</sup>: 3550, 1775, 1725, 1630, 1250, 1025.

Fractions 254–272 (500 mg) were combined and purified by prep. TLC (CHCl<sub>3</sub>-MeOH, 4:1). This gave 80 mg 2b as a gum; IR  $v_{\rm max}^{\rm Film}$  cm<sup>-1</sup>: 3500, 1770, 1050. The low resolution MS did not exhibit the molecular ion. Acetylation of 10 mg 2b

(Ac<sub>2</sub>O-pyridine overnight) and purification by TLC furnished 10 mg of the penta-acetate, **2d**; IR v<sub>max</sub><sup>CHCl<sub>3</sub></sup>cm<sup>-1</sup>: 1775, 1750-1720 (broad band). The <sup>1</sup>H NMR spectrum of **2b** and the <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra of **2d** are listed in Tables 1 and 2.

Enzymatic hydrolysis of 2a. A mixture of 28 mg 2a in 1.5 ml  $\rm H_2O$  and 18 mg  $\beta$ -D-glucosidase was stirred for 12 hr at room temp. during which time TLC indicated disappearance of starting material. The mixture was diluted with 30 ml  $\rm H_2O$  and extracted with EtOAc (3 × 100 ml). Evaporation of the washed and dried extract gave 15 mg 2a identical (TLC, IR, NMR, MS) with authentic material. The aq. portion furnished glucose.

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#### REFERENCES

- Wagner, H. (1977) in The Biology and Chemistry of the Compositae (Heywood, V. H., Harborne, J. B. and Turner, B. L., eds.) Academic Press, London. p. 1017.
- 2. Klein, E. and Thömel, F. (1976) Tetrahedron 32, 163.
- Govindan, S. and Bhattacharyya, S. C. (1977) Indian J. Chem. 15B, 956.
- Chugunov, P. V., Rybalko, K. S. and Shreter, A. I. (1971) Khim. Prir. Soedin. 7, 727.
- Rybalko, K. S., Konovalova, O. A., Orishenko, N. D. and Shreter, A. I. (1976) Rastit. Resur. 12, 387.
- Shamyanov, I. D., Mallabaev, A. and Sidyakin, G. P. (1978) Khim. Prir. Soedin. 14, 442.
- Shamyanov, I. D., Mallabaev, A. and Sidyakin, G. P. (1979) Khim. Prir. Soedin. 15, 865.
- Shamyanov, I. D. and Sidyakin, G. P. (1980) Khim Prir. Soedin. 16, 258.
- Konovalova, O. A., Rybalko, K. S. and Pimenov, M. G. (1979) Khim. Prir. Soedin. 15, 865.
- Singhal, A. K., Chowdhury, P. K., Sharma, R. P., Baruah, J. N. and Herz, W. (1982) Phytochemistry 21, 462.
- Gonzalez, A. G., Bermejo, J., Cabrera, I., Massanet, G. M., Mansilla, H. and Galindo, A. (1978) Phytochemistry 17, 955.
- Rustayian, A., Niknejad, A., Zdero, C. and Bohlmann, F. (1981) Phytochemistry 20, 2427.
- 13. Gonzalez, A. G., de la Rosa, A. D. and Massanet, G. M. (1982) *Phytochemistry* 21, 895.
- 14. Stevens, K. L. (1982) Phytochemistry 21, 1093.
- Rustayian, A., Niknejad, A., Sigan, H. and Ahmad, A. (1981) Fitoterapia 52, 31.
- Fischer, N. H., Olivier, E. J. and Fischer, H. D. (1979) Prog. Chem. Nat. Prod. 38, 47.
- 17. Zakirov, S. Kh., Kasymov, Sh. Z., Abdullaev, N. D. and Sidyakin, G. P. (1975) Khim. Prir. Soedin. 11, 261.
- Bohlmann, F. and Chen, Z.-L. (1982) Phytochemistry 21, 2120.
- Samek, Z. and Buděšinsky, M. (1979) Collect. Czech. Chem. Commun. 44, 559.
- Bermejo, J., Betancor, C., Breton, J. L. and Gonzalez, A. G. (1969) An. Quim. 65, 285.
- Gonzalez, A. G., Garcia Marrero, B. and Breton, J. L. (1970) Analyt. Chem. 66, 799.
- 22. Gonzalez, A. G. and Marrero, B. G. (1978) An Quim. 74, 1121.
- Zakirov, S. Kh., Kasymov, Sh. Z. and Sidyakin, G. P. (1976) Khim. Prir. Soedin. 12, 398.
- Narayanan, C. R. and Venkatasubramanian, N. K. (1968) J. Org. Chem. 33, 3156.

<sup>†</sup>Assignment by selective spin decoupling.